

# Pogil Significant Zeros Answer Key

## Chapter 1 : Pogil Significant Zeros Answer Key

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Significant digits and measurement pogil key page 1 1) zero and 10 cm 2) no, they were not. 3) students might have split the ruler in half, then in half again or they might have split the ruler in thirds. page 2 4) whole numbers between 1 and 10 cm (0, 1,2,3,4,5,6,7,8,9, and 10 cm or 0-10 cm) 5) yes! the 3 (or the one's place)

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### [PDF] Pogil Significant Zeros Temecula Valley Unified School

Give two examples to support your answer. 23. write each of the measurements below in expanded notation and underline the significant digits. a.  $5.0780 \times 10^6$  g = b.  $4.800 \times 10^4$  l = c.  $0.7200 \times 10^4$  mm = d.  $3700 \times 10^3$  cm = title: pogil significant zeros author:

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All of the answers can be found online for almost every single pogil..... currently, there is a significant amount of discussion on teaching list serves about the frustration of people posting answer keys online and students checking the internet instead of doing the work. many teachers, myself included, experience this frustration.

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## [PDF] Significant Zeros Bohs Cp Chemistry Home

If you removed a significant zero from the end of a number, would the numeric value of the number change? \_\_\_\_\_ ... adapted from pogil activities for high school chemistry . ... give two examples to support your answer ...

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## [PDF] Units Measurement Uncertainty And Significant Figures

Because the two lengths were quoted to two significant digits, the answer should have only two significant digits; i.e., 0.56 m<sup>2</sup>. 12. a one-gram standard reference weight is repeatedly weighed on an analytical balance. the readings from the balance are as follows: 1.003 g, 0.9998 g, 1.005 g, 0.9995 g. comment on the precision and accuracy of these data.

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## [PDF] Sig Figs Homework Problem Answer Key 3 0800 Five

Rule one " the 3 and the 8. rule two " the zero between the 3 and 8. rule three " the two trailing zeros after the 8  
2) 0.00418 - three significant figures: the 4, the 1, and the 8. this is a typical type of problem where the student errs by giving five significant figures as the answer. 3)  $7.09 \times 10^{-5}$  - three significant figures.

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## [PDF] Chem 115 Pogil Worksheet Week 1 Units Measurement

Significant figures of a number and to establish the correct number of significant figures in the answer to a calculation.

(1) for decimal numbers with absolute value >1, all digits are significant. 2.620 4 sig. figs. 50.003 5 sig. figs. (2) if there is no decimal point, zeroes that set magnitude only are not significant. 103,000 3 sig. figs.

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## [PDF] Why Model 1 Mass Of Rocks Quia

Rule 1: all non-zero numbers are significant. rule 2: sandwiched zeros (those that occur between two significant digits) are significant. rule 3: zeros that are only placeholders for a decimal are not significant. rule 4: zeros at the end of a number that also contains a decimal are significant.

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